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AUTHOR(S):

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GOTOH LABORATORY (July 1947~)

Head: Dr. Rempei Gotoh

Prof. Rempei Gotoh succeeded Prof. Shinkichi Horiba in 1947. Scientific works carried out in this laboratory before 1951 from that time were reviewed in the Commemoration Volume for the Silber Jubilee of the Institute for Chemical Research. Main subjects of the research since 1951 are as follows:

1. Dielectric properties of pure and disperse systems.
2. Infrared spectroscopy of polar organic substances.
3. Surface and colloid chemistry.
4. Rheology.

Papers published during these fifteen years are reviewed briefly in the following.

I. Dielectric Properties of Pure and Disperse Systems

1. Spherical disperse system-emulsions: Dielectric properties of emulsions have been extensively studied. In oil-in-water-emulsions no dielectric dispersion was found and Bruggeman's equation holds for both the dielectric constant and the conductivity of the system. With water-in-oil emulsions, however, a remarkable dielectric dispersion occurred and the limiting dielectric constant at high frequency was fitted by Bruggeman's equation. Bruggeman's equation was extended to a general spherical disperse system in which the disperse phase and medium have different dielectric constants and conductivities. The new equation derived from the theory of interfacial polarization in spherical disperse system holds for the dielectric relaxation as well as the static dielectric properties of a variety of emulsion over the whole concentration range of disperse phase.

2. Suspensions: Dielectric properties of solid powders dispersed in liquid paraffin were studied. Unlike the results on emulsions, the dielectric dispersion due to the interfacial polarization was not found, but moisture adsorbed on solid powder particles contributed to a relaxation of the wedge type.

3. Binary mixtures: In hexane-nitrobenzene mixture either an emulsion or a molecular solution was realized under the condition of composition of a mixture and temperature. The dielectric constants of this particular system were discussed, based on dielectric theories of a molecular solutions and a disperse system.

4. Polar Liquids: Dielectric properties of pure low-molecular weight polyethylene glycols were investigated over a wide range of frequency. The static dielectric constants of the glycols showed typical dielectric behavior of hydrogen-bonded liquids. The dielectric relaxation in the glycols was observed at microwave frequencies. A striking feature in the relation between relaxation time and the chain length was discussed in terms of flexibility of the molecular chain.

II. Infrared Spectroscopy of Polar Organic Substances

(1) Surface activities and molecular Structures

(A) Oxydation of liquid paraffin: Infrared absorption of liquid paraffin heated in the atomosphere of oxygen and nitrogen were measured at various temperatures. While vinyl compounds were decteted in nitrogen at above 300°C, formation of OH and CO radicals appeared in oxygen at above 150°C. Surface activities were observed only for the specimens heated in oxygen atomsphere.

(B) Inductive effect of polar substituents on CH stretching vibration: Infrared absorption studies were carried out of the CH stretching frequencies for homologous series of aliphatic alcohols, fatty acids, alkyl bromides as well as n-hydrocarbons. It was pointed out that the methyl stretching frequencies ν decrease with the carbon chain length, n, following the relation;

$$\nu - \nu_0 = Ae^{-Bn},$$

where ν_0 is the methyl frequency of saturated hydrocarbon and both A and B are experimental constants. Above relation was derived theoretically and interpreted in terms of inductive effect of electronegativities of ploar substituents.

(2) Application of polarized infrared spectroscopy: Molecular structure and orientation of organic substances were investigated by means of polarized infrared spectroscopy. A new method was proposed for the determination of the direction of the transition moments with a given molecular vibration, and applied to molecular crystals with C_s , S_{2h} or C_2 symmetry. A differential method of polarized infrared spectroscopy was devised and applied to the study of orientation of polymer molecules such as vulcanized natural rubber, polyethyleneand polyvinyl chloride during elongation as well as stress relaxation.

III. Surface and Colloid Chemistry

1 Interaction between dyestuffs and surfactants: Chromatography and capillary analysis were applied to the study of the interaction between dyestuffs and surfactants, especcially non-ionic surfactants. A theory of capillary analysis was proposed and formation of complexes of dyes and surfactant as well as association of dyes and surfactants were discussed. Interaction between dyes and DNA (deoxyribonucleic acid) were also discussed.

2 Electrocapillary phenomena:

(1) Theory of coagulation of colloids and coalescence of mercury droplets: The potential ranges of polarization of a pair of mercury droplets, over which the two droplets coalesced, were measured as functions of ionic concentrations for various electrolytes. The condition of coalescence was proved to be in quantitative agreement with the Verwey-Overbeek theory of coagulation of hydrophobic colloids.

(2) Electrocapillary curves of oil-water interfaces: Electrocapillary curves were obtained for the systems of organic solvents in contact with aqueous solutions.

The electrocapillary curve depended upon the ionic type of the surface active agents in the oil phase. The experimental results were discussed with respect to the ionic structure of the electric double layer at the oil-water interface.

IV. Rheology

1 Viscoelastic behaviors of colloidal materials:

Viscoelasticities of asphalts and bentonite gels were measured by means of a cone and plate type viscometer. Viscosity coefficients, delayed elastic compliance and the retardation spectra of asphalts were obtained from the strain-time and recovery curves at various temperatures. Bentonite gels showed viscoelastic behavior between two limits of shearing stress. Characteristic slip fracture were observed above critical shearing stress, which depended upon the concentration of bentonite.

2 Mechanical fracture of colloidal materials:

(1) Brittle velocity: Type of mechanical fracture or breakdown of viscous materials such as rosin, asphalts and plastisized polymers were examined at various rate of extension and temperatures. It was found that the critical velocity for brittle fracture and for dropping break-down increased exponentially with increasing temperature. Creeping or spinnable extension was observed between those two limiting conditions.

(2) Slip fracture and shear strength of bentonite gels: Shear modulus and shear strength for slip fracture were measured by various methods over the wide range of concentration of bentonite. Shear modulus, G , showed characteristic dependency on the concentration similar to those of gel elasticity of high polymers. It was found that the shear strength F_0 was related to the shear strength as follows:

$$F_0 = \nu_0 G,$$

where ν_0 is the critical strain and took almost theoretical values ranging 1/10—1/30.

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